# Part A: Buffer and standard solution preparation

## 1) Phosphate buffer (PB) 0.1 M pH = 7 preparation

Calculate the quantity of PB (MW = 119.98 g/mol) to be weighted in order to prepare a 100 mL solution of PB 0.1 M. Correct the pH to 7 by using NaOH 5 M. Release of NaOH 'drop-by-drop' till the pH = 7 is reached.

## 2) Mother solutions of Caffeic Acid (CF) and p-Coumaric Acid (CM) preparation

Calculate the quantity of CF (MW = 180.16 g/mol) and CU (MW = 164.05 g/mol) to be weighted in order to prepare a 5 mL solution at the concentration of 20 mM in methanol for both the compounds.

## 3) Intermediate mother solutions of CF and CU preparation

Starting from the mother solutions prepared in step 2, prepare intermediate mother solutions in PB 0.1 M pH = 7 (prepared in step 1).

- CF 250 μM
- CU 250 μM
- CF 250 μM e CU 500 μM

# **Part B: Samples preparation**

- 1) Prepare a solution 80:20 (v/v) MeOH/H<sub>2</sub>O
- 2) Weight 500 mg of sample
- 3) Add 10 mL of the solution prepared in step 1
- 4) Orbital shaker agitation for 1 h (in the dark)
- 5) Filtration with the syringe by using PTFE 0.45  $\mu$ m

# Part C: Samples solution preparation for standard addition calibration

- 1) Dilute the samples:
  - Oregano: 1.10000
  - Parsley: 1.1000
  - Mint: 1.1000
- 2) Calculate the volume of standard solutions to be added to each sample.

NB: pay attention to the total volume reached in order to calculate the exact concentration of the analytes!

#### Oregano

STANDARD ADDITION			Diluted sample	CF + CU	Calculate the final volume
	CF (µM)	CU (µM)	(µL)	(µL)	(µL)
0 (sample)	-	-	1000		
<b>1</b> °	2.5	5	1000		
<b>2</b> °	5	10	1000		
<b>3</b> °	7.5	15	1000		

## Parsley

STANDARD ADDITION			Diluted sample	CU	Calculate the final volume
	CF (µM)	CU (µM)	(μL)	(µL)	(µL)
0 (sample)	-	-	1000		
<b>1</b> °	-	2.5	1000		
<b>2</b> °	-	5	1000		
<b>3</b> °	-	7.5	1000		

#### Mint

STANDARD ADDITION			Diluted sample	CF	Calculate the final volume
	CF (µM)	CU (µM)	(µL)	(µL)	(µL)
0 (sample)	-	-	1000		
<b>1</b> °	2.5	-	1000		
<b>2</b> °	5	-	1000		
<b>3</b> °	7.5	-	1000		

#### Sage

STANDARD ADDITION			Diluted sample	CF	Calculate the final volume
	CF (µM)	CU (µM)	(µL)	(µL)	(µL)
0 (sample)	-	-	1000		
<b>1</b> °	2.5	-	1000		
<b>2</b> °	5	-	1000		
<b>3</b> °	7.5	-	1000		

# Part D: Running DPV measures

Drop 80  $\mu\text{L}$  of the solution onto the electrode surface. Ensure to cover the whole electrode.

Record the peak values (current) obtained for CF and CU measurements.

Parameters

- t equilibration: 5s
- E begin: -0.35 V
- E end: 0.6V
- E step: 0.0025V
- E pulse: 0.05V
- t pulse: 0.05s
- scan rate: 0.025 V/s

#### Oregano

STANDARD ADDITION			CF-like peak	CU-like peak
	CF (µM)	CU (µM)	(µA)	(µA)
0 (sample)	-	-		
1°	2.5	5		
2°	5	10		
3°	7.5	15		

## Parsley

STANDARD ADDITION			CU-like peak
	CF (µM)	CU (µM)	(µA)
0 (sample)	-	-	
1°	-	2.5	
2°	-	5	
3°	-	7.5	

## Mint

STAN	CF-like peak		
	CF (µM)	CU (µM)	(μΑ)
0 (sample)	-	-	
1°	2.5	-	
2°	5	-	
3°	7.5	-	

## Sage

STAN	DARD ADDI	CF-like peak	
	CF (µM)	CU (µM)	(µA)
0 (sample)	-	-	
1°	2.5	-	
2°	5	-	
3°	7.5	-	

# Part E: Building of standard addition plot

- Plot the height of the peaks vs. [μM] of the standards for all three components, including the unfortified sample (remember to use the final concentration of the standard in the final volume)
- 2) Determine the least-squares value of the resulting plots, and record both the slope and intercept.
- 3) Determine the  $[\mu M]$  of the **CF and CU equivalents** in the unknown sample from the equation:  $[\mu M] = -q/m$  (put y=0). Pay attention to the dilution factors.