## **Electronic noses**



environmental monitoring for air quality control



food quality and safety control



cosmetic industry control





'An electronic nose is an instrument, which comprises an array of electronic chemical sensors with partial specificity and an appropriate patternrecognition system, capable of recognizing simple or complex odors' [1]

Electronic nose

[1] Gardner, Julian W., and Philip N. Bartlett. "A brief history of electronic noses." Sensors and Actuators B: Chemical 18.1-3 (1994): 210-211.



MOS SENSORS

Metal oxide semiconductors



Constituted by three main parts:

- Ceramic substrate
- •Heating wire or thermistor
- Semiconducting metal oxides film (Zn, Co, etc.)

They measure conductivity changes onto the surface of the sensors induced by gases. Sensitive to combustion gases (hydrocarbons, NO, CO). Work at 300-400°C. An exchange between the gas and the oxygen on the film causes a change in resistance dedendent on the adsorbed gas.

# MOSFET SENSORS

## Field effect metal oxides transistors

Made by 3 parts:

- Semiconducting Silicon
- Insulating silica layer
- Catalytic metal (Pt, Pd, etc.)



Work as a transistor at applied potential at 140-170°C. Sensitive to compounds containing hydrogen (amines, aldehydes, esters, chetons, aromatics ed alcohols). Whwn a polar molecules interacts with the metal the electric field is modified and a change in current occurs. The device output is the voltage necessary to have the current back at the initial value.

## QUARTZ CRYSTAL MICRO-BALANCE



7 mm gold

## Resonant frequency 20 MHz



### Chocolate

- **⊘** Temperature: 40°C
- Equilibration time: 10 min
- ⊘ 15g in 100 mL lab bottle grated and melted
  ⊘ 4 L/h

## **Standard Samples**

## Off-flavoured samples

## **PLS-DA** analysis

Off-Flavour	Process
3 methylbutanal	
Phenylacetaldehyde	Fermentation volatiles
Acetic Acid	Conching process
Tetramethylpyrazine	
2-acetylpyrrole	Roasting Process
2-nonenal	
2,4-decadienal (t,t)	Fat related (oxidation)









Off-flavours were preliminarily added in the cocoa butter to achieve the concentration of 125 ppm. One tea spoon of contaminated cocoa butter was then added to 400 g of chocolate to obtain an estimated final concentration in the sample of ~ 6ppm.



## ELECTRONIC NOSE SENSOR ARRAYS

## ✓✓ GNP-GLUTATHIONE

GNP-Peptide based

- **⊘ GNP-CYS-GLY GNP-CYS**
- GNP-THIOGLICOLIC ACID GNP-
- **CYS-ARG-GLN-VAL-PHE GNP-**
- **⊘** CYS-ILE-HIS-ASN-PRO GNP-CYS-
- ✓ ILE-GLN-PRO-VAL GNP

 $\bigcirc$ 



- CU-BUTI-TPP
- ✓ CO-BUTI-TPP
- **ZN-BUTI-TPP**
- **MN-BUTI-TPP**
- **FE-BUTI-TPP SN-**
- **BUTI-TPP** H<sub>2</sub>-
- **BUTI-TPP**

 $\bigcirc$ 

MG-BUTI-TPP

### Dark Chocolate

### White Chocolate



### AuNP-Peptide vs. Porphyrin

### GNP-Peptide based

UNITE

	Regular	Off Flavours	% sCorrect		Regular	Off Flavours	% sCorrect		Regular	Off Flavours	% S <b>Correct</b>
Regular	48	0	100	Regular	39	0	100	Regular	51		98
Off flavours	0	7	100	Off flavours	0	7	100	Off flavours	0	7	100
		Tot. Corr	ect:			Tot. Corr	ect:			Tot. Cor	rect:
	Regular	Off Flavours	% sCorrect		Regular	Off Flavours	% ©Correct		Regular	Off Flavours	% Correct
Regular	14		93	Regular	13		92	Regular	15		94
Off flavours		9	90	Off flavours	4	8	67	Off flavours	4	8	67
		Tot. Cor	rect:				Tot. Correct:				
	D Comp	aunone	M Faieta I	n Pizzoni	C Di Nata	ale et al .	Sensors a	nd Actual	<i>tor R</i> 207	1114 2015	i



### ELECTRONIC NOSE SEQUENCES...



#### **Materials and Methods**



### **Materials and Methods**

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<u>SPME</u> <u>ANA</u>	<u>-GC/MS</u> <u>LYSIS</u>	Confirmed by Retention index  Confirmed by Retention  Confirmed  Confirmed by Retention  Confirmed	lex
headspace analysis	Desorption	Image: Note of the second s	
		Preside    Ref <sup>and</sup> Matabolis    Matabolis    Matabolis    Matabolis    Matabolis    Personal matabolis    Persona	

a) Retention time (min).

b) Kovat's retention index reported in the literature for BP-20 capillary column or equivalents [53]. c) Kovat's retention index relative n-alkanes( $C_8$ – $C_{20}$ ) on a BP-20 capillary column.



Example of frequency shift of four sequences and 1:9 v/v and 2:8 v/v water/solvent mixtures. A, A' =water/acetonitrile; B,B'=water/methanol; C,C'=water/acetone (yellow = glutathione, grey=TGKF, orange=IHRI, light blue=LGFD.

Example of frequency shift of the ZnONPs sensor array with 1:9 v/v and 3:7 v/v water/solvent mixtures. A, A' =water/acetonitrile; B, B'=water/methanol. (Yellow=TGKFC, Light blue=IHRIC, blue=LGFDC, brown=KSDSC; orange=glutathione; green=WHVS; grey=ZnONPs ).



PCA of the piezoelectric response. The biplot (Score and loading) of the first two principal components showed 94.45% of the cumulative variance. Rows normalization were applied to the gas sensors array dataset. Data were auto scaled before PCA.



(A) The biplot values (sores and loadings) for fruit juices analysis along with the percentage (B) of fruit quantity and sugar concentration in the different fruit juices.



- Drying is the most important unit operation in pasta;
- the high-temperature (HT) drying technology has been widely applied by pasta manufacturers;
- HT drying has a positive influence on the mechanical properties of pasta;
- To avoid Maillard's reaction as much as possible, it is necessary to monitor volatile compounds, which may be markers of the quality of the finished product.

Volatile compounds																		
	A	Α	Α	В	В	В	С	С	С	D	D	D	Е	Е	Е	F	F	F
3-methyl-furan	-	-	-	-	-	-	-	-	1±1.15	-	-	-	-	-	-	-	-	-
pentanal	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
lattamide	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
1,3-butanediol	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
2-furanmethanol	-	-	-	-	-	-	-	-	-	1±0.58	-	-	-	7±0.6	-	-	-	-
furfuryl alcohol	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
hexanal	24±1.5	30±2.65	23±1.53	15±2.08	41±4.04	7±2.00	-	2±2.0 8	-	-	2±2.31	11±1.00	3±2.52	-	1±1.53	3±1.15	1±0.58	1±0.58
2-hexenal	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
benzaldehyde	1±1.15	3±1.00	-	2±2.08	7±2.00	2±1.52	-	-	-	-	-	-	-	-	-	-	-	-
4-methylbenzaldehyde	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
3-methylbenzaldehyde	-	-	-	-	-	-	-	-	-	3±1.00	-	-	-	10±1.52	6±1.00	4±0.57	5±1.15	6±1.00
2-pentylfuran	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
nonanal	6±1.00	3±1.52	13±1.53	15±4.50	10±1.00	5±4.00	2±1. 52	2±1.1 5	4±2.00	4±1.52	1±1.15	5±2.00	4±1.52	-	-	-	-	-
2-nonenal	10±2.00	7±0.57	4±3.51	5±3.00	1±1.00	-	-	-	-	-	-	1±1.52	-	-	-	-	-	-
decanal	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	7±4.5
3,7-dimethyl-1-octanol	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
2,2,4,4-															1+1 50	1+2.00	1+1 22	15±3.5
tetramethyltetrahydrofuran	-	-	-	-	-	-	-	-	-	-	-	-	-	-	111.55	112.00	111.33	1
2-isopropyl-5-methyl-1-	-	-	-	-	-	-	-	-	-	-	-	-	3±3.00	6±0.57	6±1.18	9±3.00	10±3.08	14±3.0 5
neptanol		_	_	_	_	_	_	_	_	_	_	_	_	_	_	_	_	-
2-butyl-1-octanol	-	-	-	- 1+1 52	-	-	-	-	-	-	-	- 1+1 15	-	-	-	-	-	-
1,1-dodecanediol	-	1±0.59	-	111.03	2±3.05	-	-	-	-	-	-	111.15	-	-	-	-	-	-

Volatile compounds of high-price pasta samples. Data were expressed as % of the total GC area.

Volatile Compounds		WHVSC	IHRIC	TGKFC	KSDSC	LAWHC	LGFDC			
Furfuryl alcohol	Mean	10	11	17	21	20	10			
	St.dev.	2	3	3	5	3	2			
	RSD-	15	25	18	22	17	15			
1,3-butanediol	Mean	9	17	21	20	16	10			
	St.dev.	1	2	4	4	4	2			
	RSD-	14	13	21	19	23	18			
benzaldehyde	Mean	17	32	54	38	26	17			
	St.dev.	2	3	4	9	6	2			
	RSD-	12	8	8	24	21	9			
2-pentylfuran	Mean	2	3	3	8	2	3			
	St.dev.	1	1	1	2	1	0			
	RSD-	37	39	41	29	35	0			
pentanal	Mean	14	36	62	41	36	15			
	St.dev.	0	1	3	3	2	1			
	RSD-	3	3	5	7	6	4			
3-methylbenzaldehyde	Mean	9	4	7	12	7	8			
	St.dev.	1	1	6	2	1	1			
	RSD-	8	35	81	18	20 9				
decanal	Mean	6	14	12	16	16	5			
	St.dev.	1	2	1	1	1	1			
	RSD-	10	11	8	4	7	11			
nonanal	Mean	8	20	14	24	23	8			
	St.dev.	1	4	2	1	2	1			
	RSD-	13	18	11	2	9	13			
2-nonenal	Mean	10	32	20	32	35	11			
	St.dev.	1	3	3	2	3	1			
	RSD-	6	11	13	5	9	5			
hexanal	Mean	15	52	36	53	55	16			
	St.dev.	2	4	3	5	2	1			
	RSD-	10	7	7	9	4	6			



Typical replicate measurement of pasta sample with ZnO-peptides based Enose. A and B were a single batch of different pasta samples at high price.



PCA of the GC-MS response for low price samples (orange) and high price samples (green). The biplot (Score and loading) of the first two principal components showed 37.08% of the cumulative variance. Data were expressed in (R.A. %) before PCA

PCA biplot (score and loading) of the normalized piezoelectric response of the gas sensors.

#### Virtual screening method



Electrostatic molecular surfaces of the ssDNA CTGCAA, with a planar interaction surface (binding score -2.26 Kcal/mol)

	CCAG	TTCT	OCOGA	TAAGT	ATAATC	CATGTC	CTGCAA
Ethanol	$-2.72 \pm 0.16$	$-2.41 \pm 0.17$	$-2.77 \pm 0.28$	$-2.28 \pm 0.23$	$-3.21 \pm 0.29$	$-3.09 \pm 0.28$	$-3.24 \pm 0.32$
3-methylbutan-1-ol	$-3.23 \pm 0.29$	$-2.77 \pm 0.17$	$-3.26 \pm 0.20$	$-2.74 \pm 0.19$	$-4.63 \pm 0.23$	$-4.12 \pm 0.29$	$-2.72 \pm 0.19$
1-pentanol	$-3.30 \pm 0.23$	$-3.22 \pm 0.32$	$-3.53 \pm 0.32$	$-2.48 \pm 0.20$	$-5.14 \pm 0.36$	$-4.21 \pm 0.42$	$-2.82 \pm 0.14$
Alcohols	$-2.94 \pm 0.15$	$-3.12 \pm 0.16$	$-3.38 \pm 0.30$	$-2.39 \pm 0.12$	$-5.55 \pm 0.28$	-4.75 ± 0.38	$-3.01 \pm 0.21$
Octanal	$-2.39 \pm 0.19$	$-3.07 \pm 0.15$	$-2.62 \pm 0.16$	$-1.54 \pm 0.08$	$-5.13 \pm 0.26$	$-3.71 \pm 0.22$	$-2.40 \pm 0.22$
Non an al	$-2.32 \pm 0.16$	$-3.24 \pm 0.26$	$-2.56 \pm 0.26$	$-1.53 \pm 0.09$	$-5.54 \pm 0.44$	$-4.05 \pm 0.36$	$-2.56 \pm 0.26$
Aldehydes	$-2.25 \pm 0.18$	$-2.78 \pm 0.28$	$-2.42 \pm 0.22$	$-1.55 \pm 0.14$	$-4.67 \pm 0.23$	$-3.39 \pm 0.17$	$-2.27 \pm 0.14$
Ethyl acetate	$-2.17 \pm 0.15$	$-2.19 \pm 0.15$	$-2.49 \pm 0.22$	$-1.51 \pm 0.14$	$-3.64 \pm 0.18$	$-2.68 \pm 0.16$	$-1.90 \pm 0.17$
Ethyl octanoate	$-1.95 \pm 0.14$	$-3.14 \pm 0.16$	$-2.49 \pm 0.17$	$-1.46 \pm 0.07$	$-6.28 \pm 0.57$	$-4.20 \pm 0.38$	$-2.26 \pm 0.11$
Esters	$-2.19 \pm 0.20$	$-2.73 \pm 0.22$	$-2.53 \pm 0.23$	$-1.58 \pm 0.16$	$-4.60 \pm 0.23$	$-3.37 \pm 0.24$	$-2.26 \pm 0.14$
Butane-2,3-dione	$-2.02 \pm 0.14$	$-2.51 \pm 0.15$	$-1.88 \pm 0.11$	$-1.44 \pm 0.09$	$-3.13 \pm 0.28$	$-2.17 \pm 0.17$	$-1.95 \pm 0.14$
Ketones	$-1.84 \pm 0.15$	$-2.17 \pm 0.11$	$-1.94 \pm 0.16$	$-1.24 \pm 0.12$	$-3.55 \pm 0.18$	$-2.43\pm0.15$	$-1.81\pm0.16$

Binding score average (Kcal/mol) of the tetramer, pentamer and hexamer DNA versus the VOCs tested in experimental part. In italic-bold, the binding score obtained by the simulations of the ssDNA versus the chemical classes (14 alcohols, 13 aldehydes, 18 esters and 5 ketones). The average and standard deviation was calculated over 10 conformers.

	CCAG Experimental K	TTCT binding ×10 <sup>3</sup> (Mol	COCGA les <sup>-1</sup> )	TAAGT	ATAATC	CATGTC	CTGCAA
Ethanol	4.3 ± 0.6	$2.3 \pm 0.1$	7.4 ± 0.5	0.9 ± 0.1	6.5 ± 0.5	8.4 ± 0.8	18.4 ± 2.8
3-methylbutan-1-ol	$6.7 \pm 0.5$	$6.5 \pm 0.6$	$23.6 \pm 3.3$	$7.0 \pm 0.6$	91.2 ± 11.9	70.1 ± 3.5	$21.7 \pm 2.2$
1-pentanol	3.3 ± 0.4	$3.1 \pm 0.3$	60.3 ± 7.8	$2.0 \pm 0.1$	$530.6 \pm 58.4$	352.8 ± 17.6	$10.3 \pm 1.3$
Octanal	$2.6 \pm 0.3$	$6.3 \pm 0.6$	4.7 ± 0.3	$2.5 \pm 0.1$	$582.2 \pm 81.5$	348.0 ± 41.8	$3.8 \pm 0.3$
Nonanal	$2.6 \pm 0.2$	7.9 ± 0.6	$6.5 \pm 0.8$	$1.9 \pm 0.1$	$283.1 \pm 14.2$	97.9 ± 8.8	$4.8 \pm 0.4$
Ethyl acetate	$0.9 \pm 0.1$	$2.4 \pm 0.3$	9.8 ± 1.5	$3.1 \pm 0.4$	$19.6 \pm 1.2$	$17.8 \pm 2.3$	$0.9 \pm 0.0$
Ethyl octanoate	$8.8 \pm 0.8$	$12.8 \pm 0.6$	43.1 ± 4.3	$2.5 \pm 0.2$	519.2 ± 41.5	345.8 ± 17.3	$6.9 \pm 0.5$
Butane - 2,3-dione	$1.4 \pm 0.2$	$3.4 \pm 0.3$	$2.9 \pm 0.3$	$3.4 \pm 0.4$	$9.2 \pm 1.1$	$4.1 \pm 0.3$	$2.5 \pm 0.4$
Correlation with Simulated results	0.16	0.63	0.65	0.37	0.83	0.67	0.80

HpDNA-AuNP sensors relative binding affinities vs the VOCs, estimated using piezoelectric response. The correlation coefficient between experimental and simulated binding is reported in the last row. The standard deviation was calculated using three measurements taken in three different days.



PCA of the piezoelectric responses of hpDNA-AuNP sensors obtained using 900 µmoles of each VOC. The plot of Scores (A) and the plot of loading (B) of the first three principal components showed 92.40% of the cumulative variance. Data were autoscaled before PCA.



- Monitoring and control of vegetable ripening are important parameters in the food industry, since the maturation state during harvest, storage and distribution on the market defines the quality of the finished product;
- To prevent enzymatic reactions during processing, storage and thawing, the packaged carrots were blanched at 95 °C for 8 min in a water bath



3 g of blanched carrots were placed in 20 ml gas-tight vials and hermetically sealed with a gas-tight septum. A total of 24 vials were prepared for each temperature tested (-18 °C, 4 °C, 25 °C, 40 °C), in order to have three replicates (three different vials) for each day of measurement (1, 4, 8, 12, 19, 26 days). Thus, a total of 140 vials for E-nose and GC-MS have been analyzed. Each vial was used only once for either the GC-MS or E-nose measurement and the relative sample was discarded.

		GC area (%)																				
										Sto	orage ti	me (days	5)									
		-18 °C					4°C						25 °C						40 °C			
Volatile compounds	1	4	8	12	19	1	4	8	12	19	26	1	4	8	12	19	26	1	4	8	12	
α-phellandrene	n.d	n.d	n.d	n.d	n.d	1	1	1	1	1	n.d	n.d**	1	n.d	n.d	1	n.d	7	1	n.d	n.d	
β-phellandrene	2	1	2	1	1	3	3	2	1	4	1	n.d	3	1	1	2	n.d	n.d	n.d	n.d	n.d	
terpinolene	n.d	n.d	1	1	0	1	1	1	n.d	1	n.d	n.d	1	n.d	n.d	1	n.d	n.d	n.d	n.d	n.d	
α-pinene	14	9	12	12	10	14	12	14	14	18	8	12	7	12	11	15	2	n.d	10	15	5	
(-) -β-pinene	3	2	3	3	3	3	3	3	3	3	2	2	3	2	2	3	1	n.d	n.d	n.d	2	
β-pinene	3	2	3	2	6	5	6	3	3	5	1	2	6	1	1	4	n.d	5	3	3	2	
Octanal	1	1	n.d	n.d	n.d	1	1	1	1	n.d	n.d	1	n.d	n.d	n.d	n.d	n.d	n.d	n.d	n.d	n.d	
γ-terpinene	7	8	9	10	7	8	9	7	8	4	1	12	7	6	5	7	n.d	21	9	9	5	
β-farnesene	1	1	1	1	1	1	1	1	1	n.d	1	1	1	1	1	1	2	n.d	n.d	n.d	n.d	
α-caryophyllene	1	1	1	1	2	2	1	1	1	1	5	1	2	1	2	1	2	n.d	n.d	n.d	n.d	
β-copaene	2	2	2	n.d	4	2	2	3	n.d	n.d	n.d	2	1	2	2	2	5	n.d	n.d	n.d	n.d	
myristicin	1	1	1	1	2	1	1	1	1	1	5	2	1	2	n.d	1	4	n.d	n.d	n.d	n.d	
elemicin	n.d	1	n.d	n.d	n.d	n.d	n.d	1	n.d	n.d	4	1	n.d	1	1	n.d	2	n.d	n.d	n.d	n.d	
butane-2,3-diol	n.d	n.d	n.d	n.d	n.d	n.d	n.d	n.d	n.d	n.d	n.d	n.d	n.d	2	2	5	13	n.d	n.d	n.d	3	
acetoin	n.d	n.d	n.d	n.d	n.d	n.d	n.d	n.d	n.d	n.d	n.d	n.d	n.d	9	3	6	15	5	3	5	6	
ethanol	n.d	n.d	n.d	n.d	n.d	n.d	n.d	n.d	n.d	n.d	n.d	n.d	n.d	n.d	n.d	n.d	n.d	5	n.d	n.d	6	
lactamide	n.d	n.d	n.d	n.d	n.d	n.d	n.d	n.d	n.d	n.d	n.d	n.d	n.d	n.d	n.d	n.d	n.d	4	11	16	7	
3-methylbutan-1-ol	n.d	n.d	n.d	n.d	n.d	n.d	n.d	n.d	n.d	n.d	n.d	n.d	n.d	n.d	n.d	1	2	1	3	3	3	

\*(mean value of n=3 repetitions); \*\*n.d.: not detected.

Results of the gas-chromatographic (GC) analysis of the headspace of carrots samples. Data are expressed as % of the total GC area\*. Frequency signal recorded with AuNPs-hpDNA testing carrots samples in triplicate



<u>GC-MS</u>

<u>E-nose</u>



PCA of the GC-MS/SPME response of carrot samples stored for different time at various temperature. Data are expressed in (Relative Abundance %) before PCA

PCA biplot (scores and loadings) of the E-nose response to carrots samples. Data were normalized before PCA