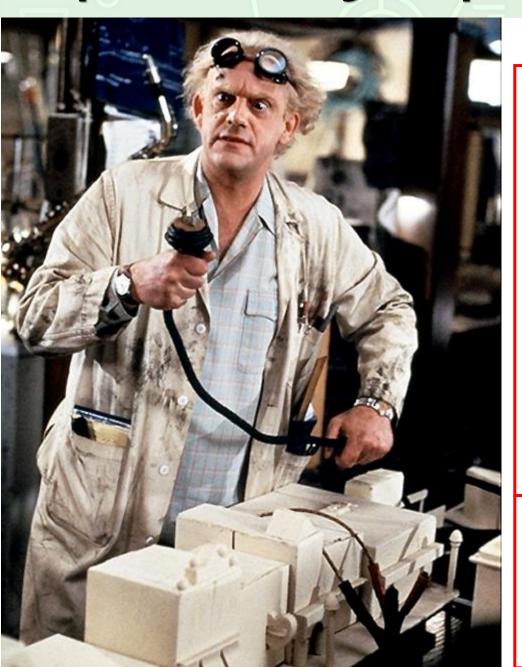


LAB-EXPERIENCE INTRODUCTION

Important knowledges/ requirment





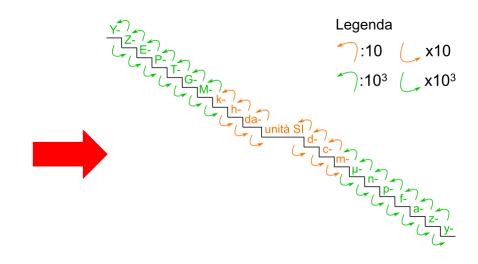
- 1- Basical knowledge(s)
- 2- Lab equipment
- 3- How to use lab equipment
- 4- Stock solution preparation
- 1) Solution preparation from a solid substrate
- 2) Solution reparation by dilution of a concentrated stock solution
- 3) Buffer solution preparation
- 4) Dilution v/v, serial dilution
 - 5- Analytes' extraction principle
 - 6- Analysis' building block(s)
 - 7- Analytical strategies

Basical knowledge(s)

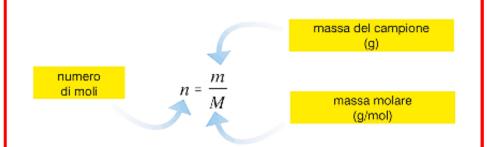


Formulary

SI Base Units				
Physical Quantity	Name of Unit	Abbreviation		
Mass	kilogram	kg		
Length	meter	m		
Time	second	S		
Temperature	kelvin	K		
Amount of substance	mole	mol		
Electric current	ampere	A		
Luminous intensity	candela	cd		



Moles and [C]



molarità = M =
$$\frac{n_{\text{soluto}} (\text{mol})}{V_{\text{soluzione}} (\text{L})}$$

The Dilution Equation

$$M_1V_1 = M_2V_2$$

 M_1 = initial molarity ("stock solution")

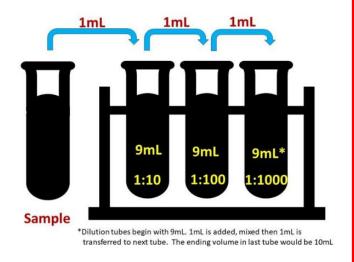
 V_1 = initial volume (Liters)

 M_2 = final (desired) molarity

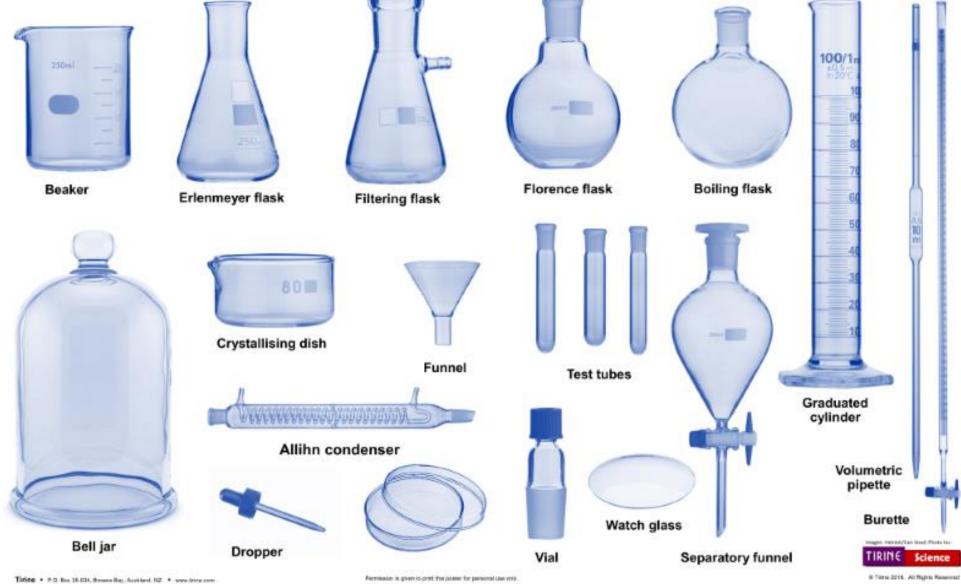
 V_2 = final volume (Liters)

This equation is used when you have a "stock solution" of higher molarity than you need and you need to dilute it to a lower molarity by adding additional solvent.

Serial dilution







Lab glassware



☐ Spray bottle





Rak







Desiccator/ Deumidificator

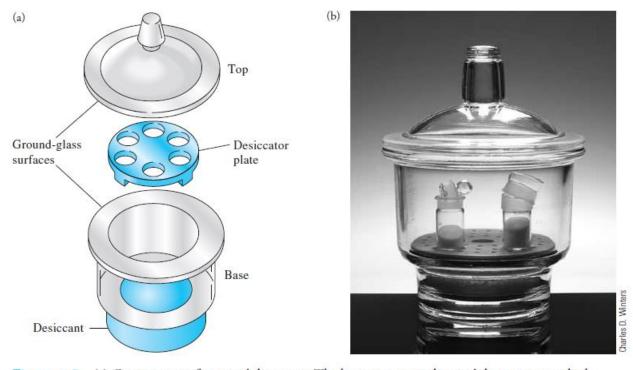
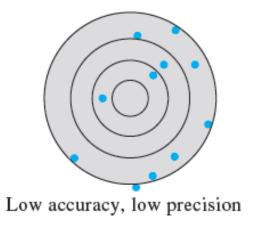
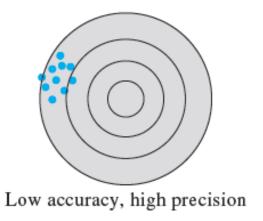


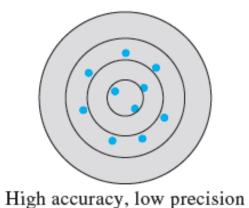
Figure 2-8 (a) Components of a typical desiccator. The base contains a chemical drying agent, which is usually covered with a wire screen and a porcelain plate with holes to accommodate weighing bottles or crucibles. (b) Photo of desiccator containing weighing bottles with dry solids.

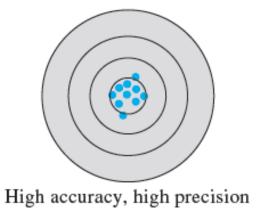
Precise or accurate?







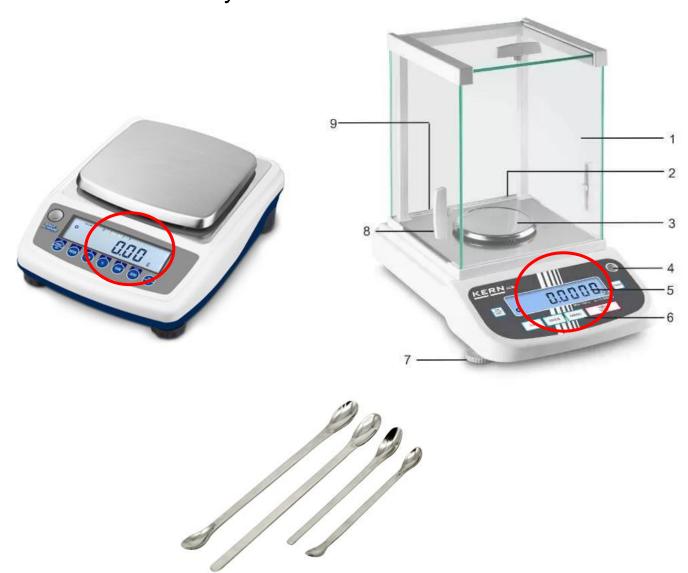




What precision means? What accuracy means?



Technical and Analytical balance





Measuring volume





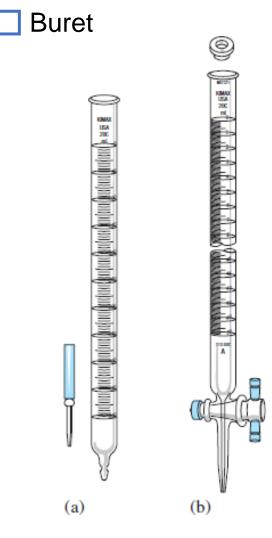


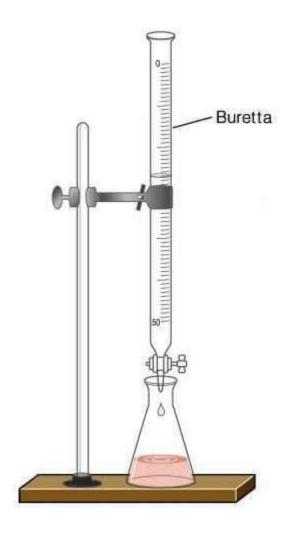
Figure 2-19 Burets:

- (a) glass-bead valve,
- (b) Teflon valve.

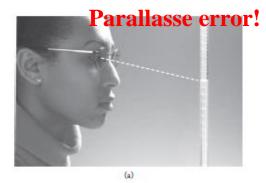
Measuring volume



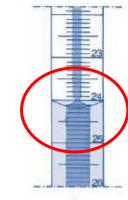




PAY ATTENTION!!!





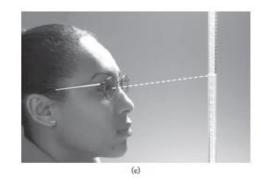














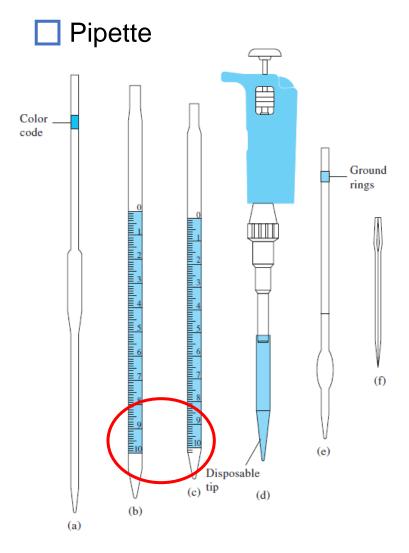


Figure 2-17 Typical pipets:
(a) volumetric pipet, (b) Mohr pipet,
(c) serological pipet, (d) Eppendorf
micropipet, (e) Ostwald–Folin pipet,
(f) lambda pipet.



Tolerances, Class A Transfer Pipets

Capacity, mL	Tolerances, mL
0.5	±0.006
1	± 0.006
2	± 0.006
5	± 0.01
10	± 0.02
20	± 0.03
25	± 0.03
50	± 0.05
100	± 0.08

TABLE 2-2

Characteristics of Pipets

Name	Type of Calibration*	Function	Available Capacity, mL	Type of Drainage
Volumetric	TD	Delivery of fixed volume	1-200	Free
Mohr	TD	Delivery of variable volume	1-25	To lower calibration line
Serological	TD	Delivery of variable volume	0.1-10	Blow out last drop**
Serological	TD	Delivery of variable volume	0.1-10	To lower calibration line
Ostwald-Folin	TD	Delivery of fixed volume	0.5-10	Blow out last drop**
Lambda	TC	Containment of fixed volume	0.001-2	Wash out with suitable solvent
Lambda	TD	Delivery of fixed volume	0.001-2	Blow out last drop**
Eppendorf	TD	Delivery of variable or fixed volume	0.001-1	Tip emptied by air displacement

^{*}TD, to deliver; TC, to contain.

^{**}A frosted ring near the top of pipets indicates that the last drop is to be blown out.

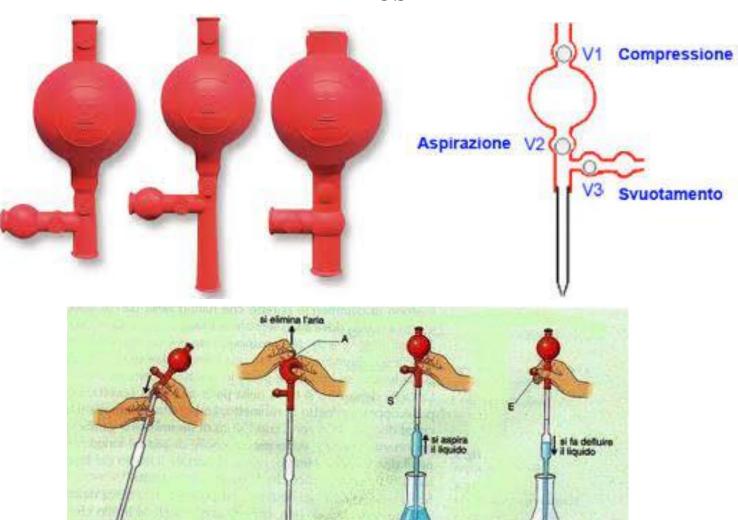
Measuring volume



☐ Glassware pipette



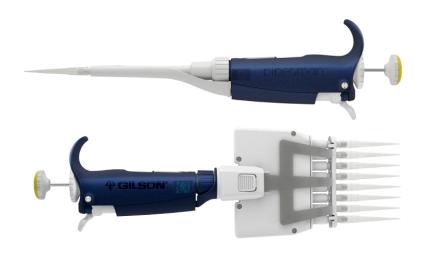
PELEUS BALL



Measuring volume

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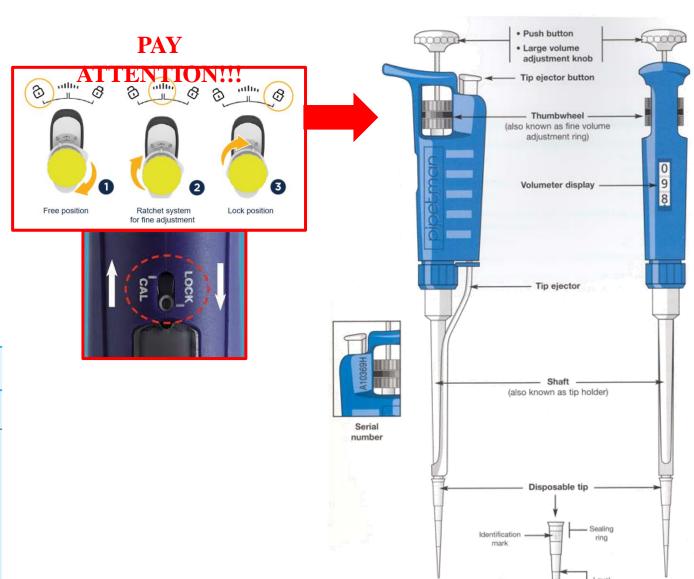
☐ Gilson pipette





Range and Precision of Typical Eppendorf Micropipets

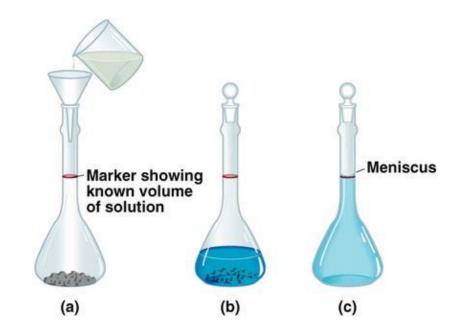
Volume	Standard
Range, µL	Deviation, µL
1–20	<0.04 @ 2 μL
	<0.06 @ 20 μL
10–100	<0.10 @ 15 μL
	<0.15 @ 100 μL
20–200	<0.15 @ 25 μL
	<0.30 @ 200 μL
100-1000	<0.6 @ 250 μL
	<1.3 @ 1000 μL
500-5000	<3 @ 1.0 mL
	<8 @ 5.0 mL

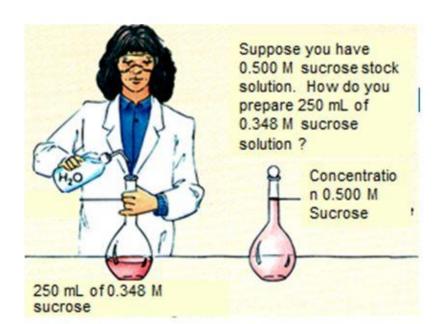




Two methods for Preparation of a desired volume of a Molar Solution

- 1) Preparation from a solid solute.
- 2) Preparation by Dilution of a Concentrated Stock Solution.



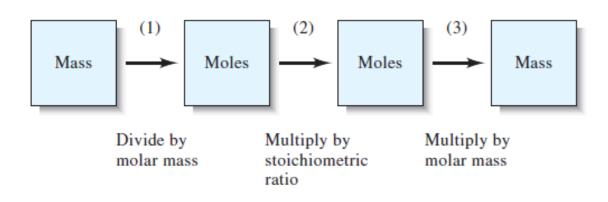


How to solve?





□ 1) Preparation from a solid substrate



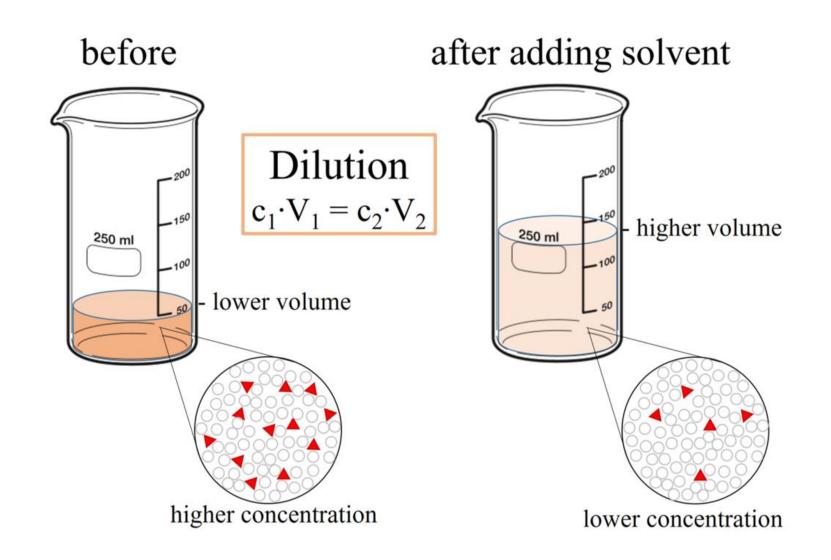




Stock solution preparation



□ 2) Preparation by dilution of a concentrated stock solution

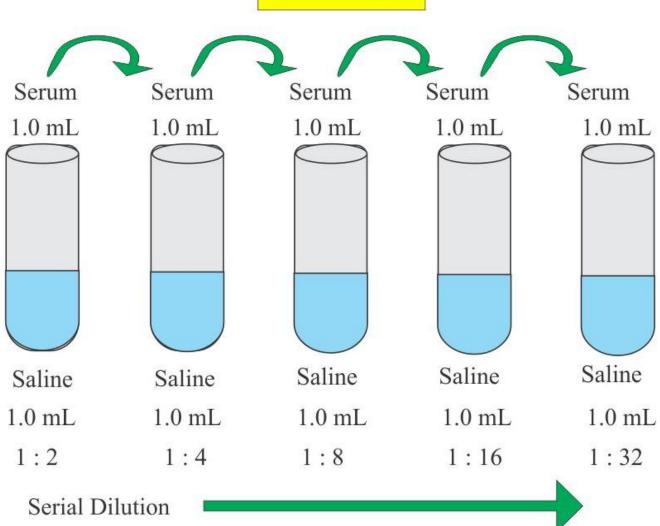


Stock solution preparation

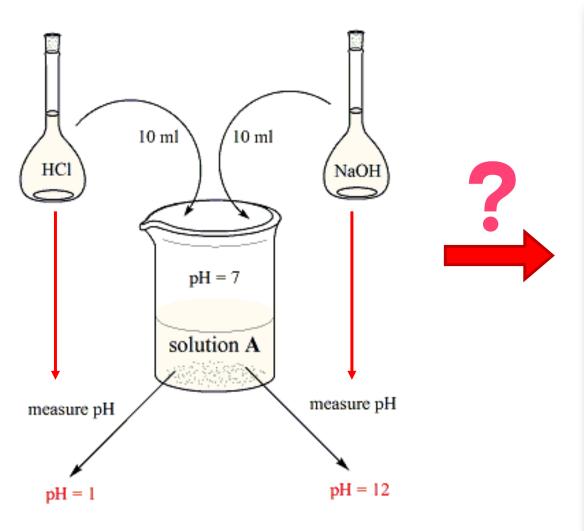


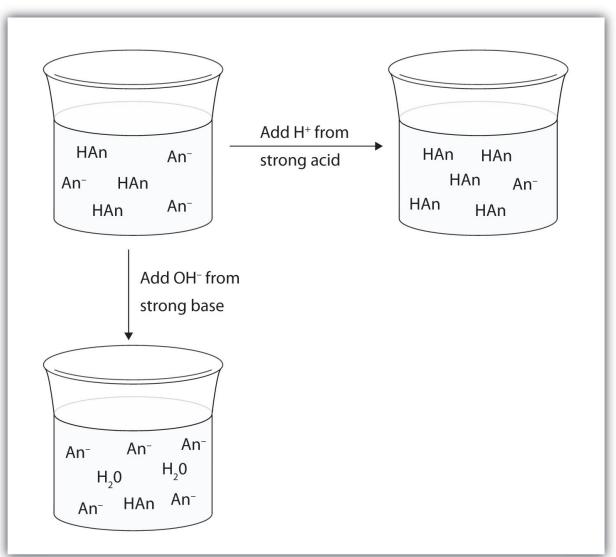
Serial dilution

Serial Dilution



□ 3) Buffer solution preparation

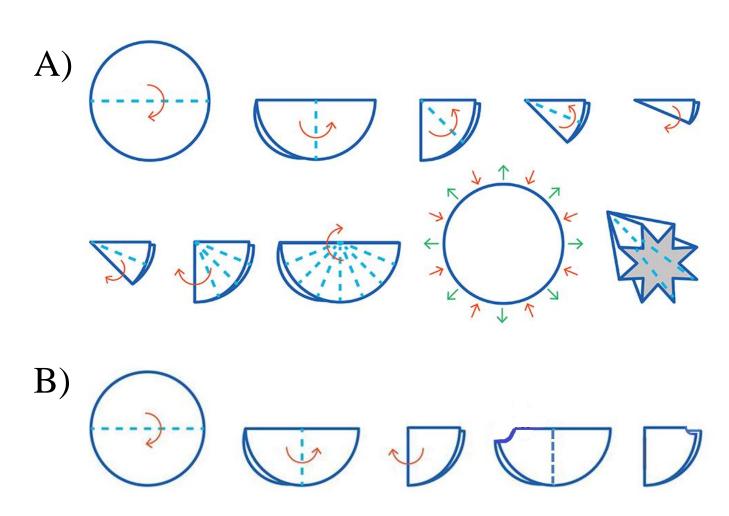


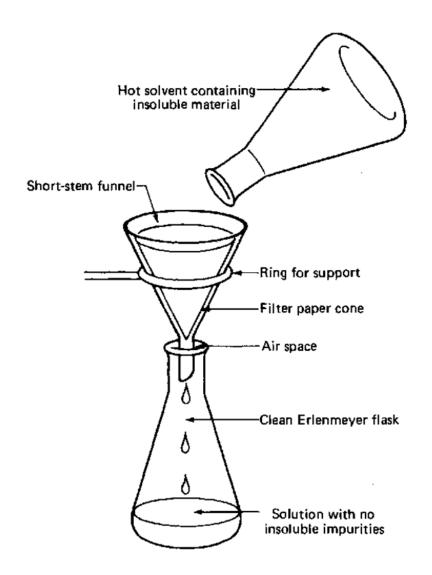


Separation system



How to build up a paper filter





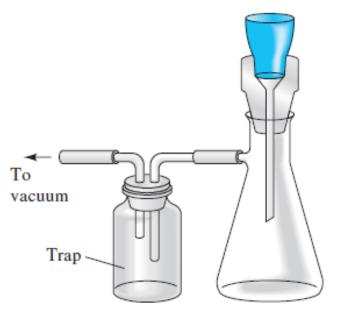
Separation system



■ Vacuum filtration

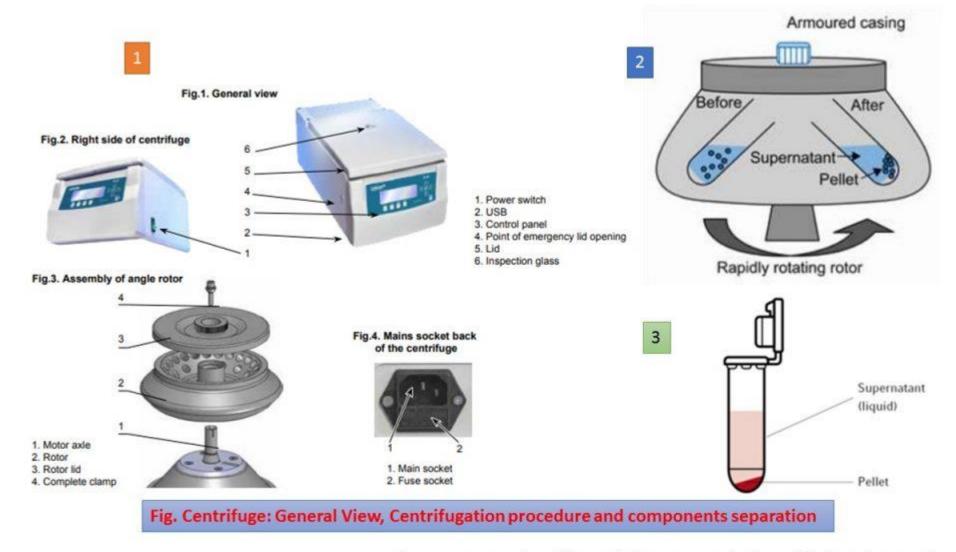
large enough particles of solid cannot fit through tiny holes in filter paper, so remain here Büchner moistened funnel filter paper porous plate (plate with holes in) rubber bung Büchner rubber tubing flask suction from aspirator creates partial vacuum in flask filtrate (liquid that passes through filter paper) collects here

Vacuum system





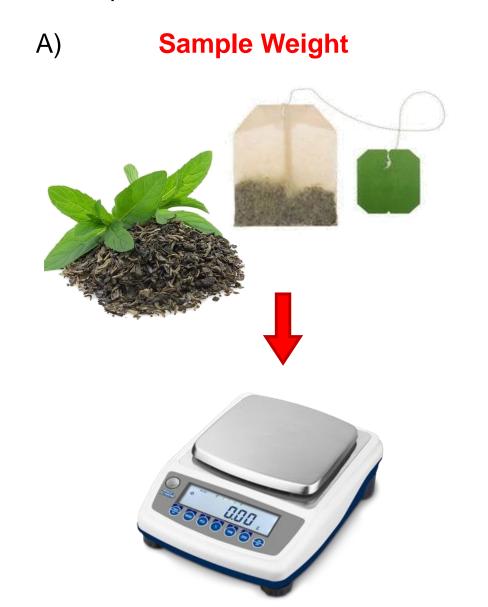
Centrifuge

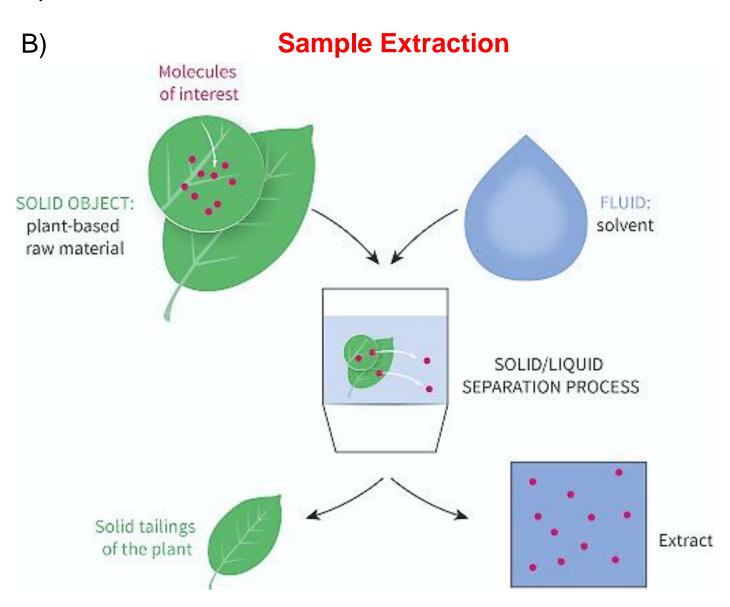


Analytes' extraction principle

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Solid-liquid extraction. Extraction form solid sample.

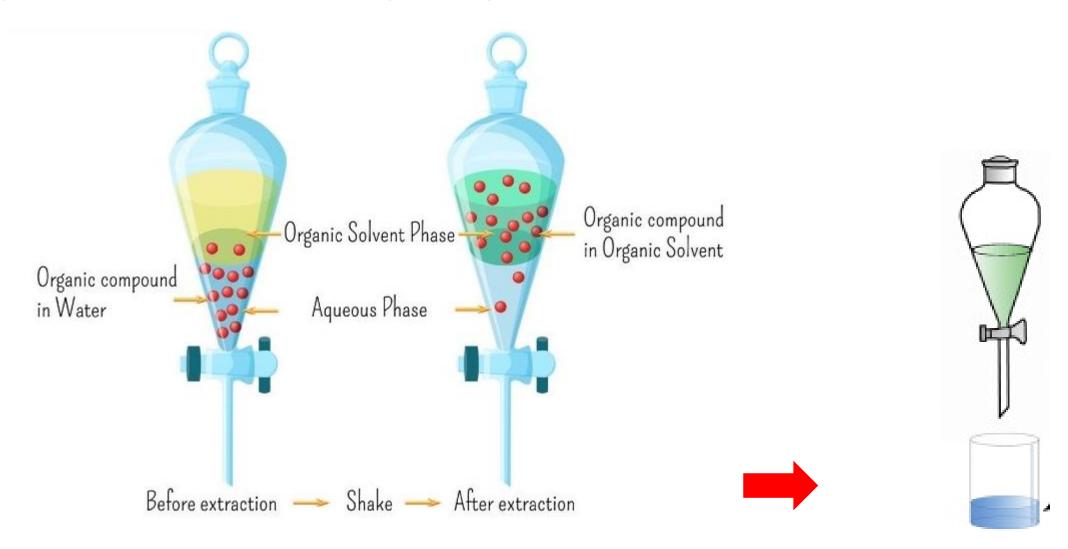




Analytes' extraction principle

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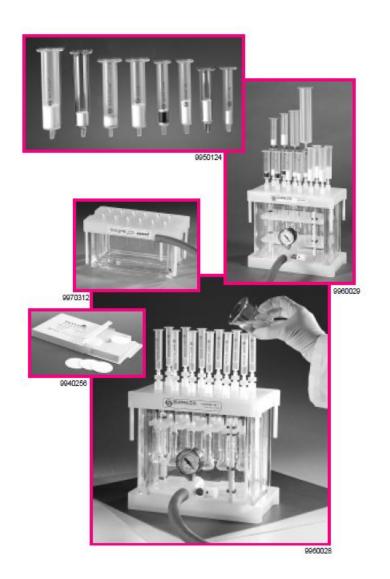
Liquid-liquid extraction. Extraction form liquid sample.

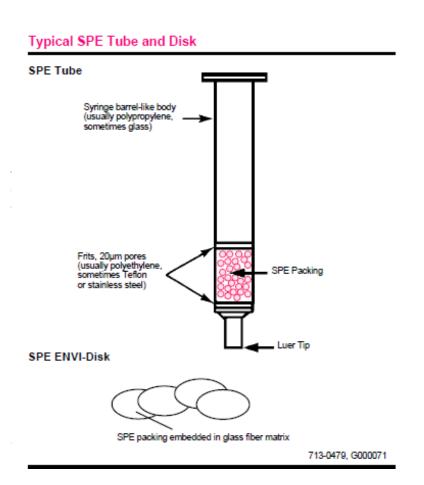


Analytes' extraction principle

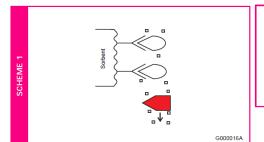
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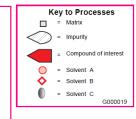
Solid phase extraction (SPE). Extraction form liquid sample.

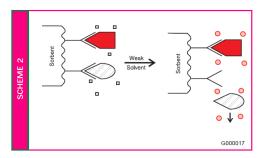


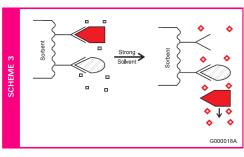


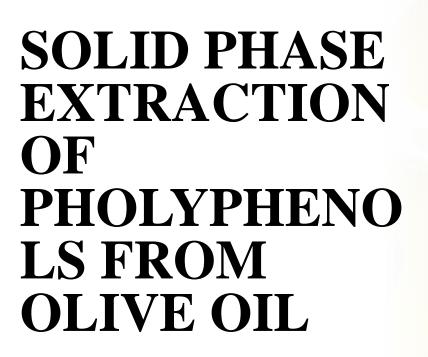
How to use SPE

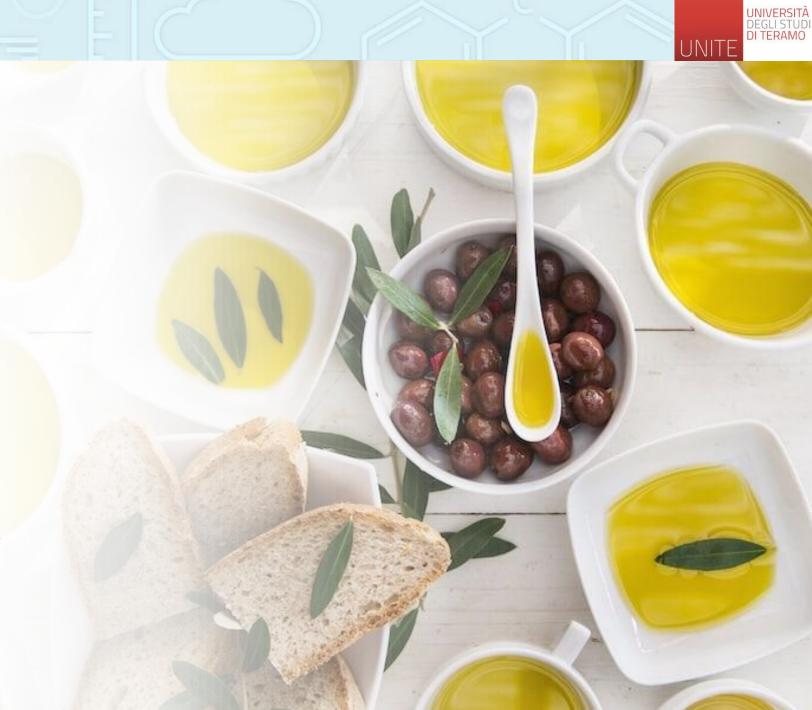














Antioxidants could be defined as sacrificial molecules.

Antioxidants are natural or synthetic molecules able to scavenge reactive species, such as reactive oxygen and nitrogen species (ROS and RNS), contributing to oxidative homeostasis.

Phenolic compounds



Beneficial effect on human health



Anti-microbial property



Additives in biomedicine practices



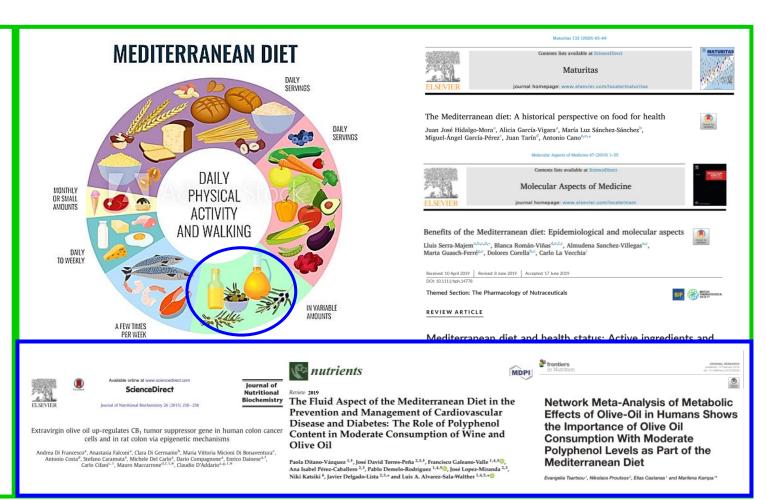
Food supplements (sensory and nutritional properties, shelf-life)



Quality and process indicators



Potential tools for functionalization of materials

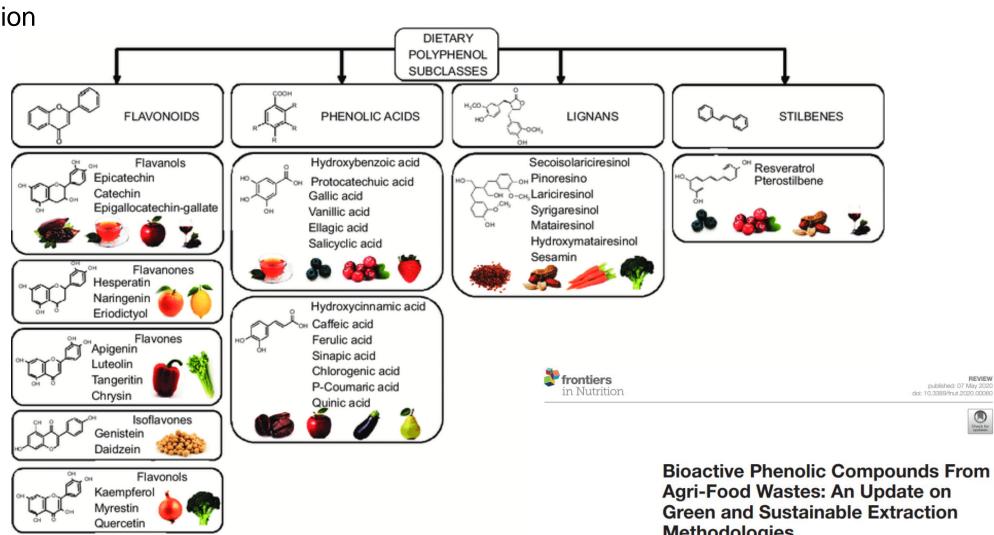


Classification

Anthocyanins

Cyanidin

Delphinidin Malvedin Pelargonidin

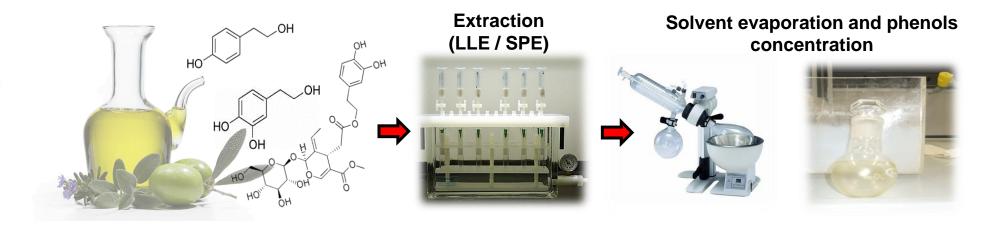


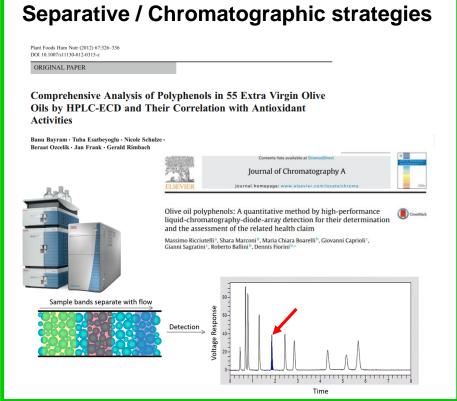
Methodologies

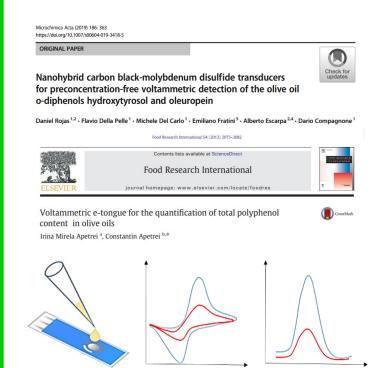
Lucia Panzella 1*, Federica Moccia 1, Rita Nasti 2, Stefania Marzorati 2, Luisella Verotta 2 and Alessandra Napolitano 1

Phenolic content evaluation in Extra Virgin Olive Oil. Main strategies

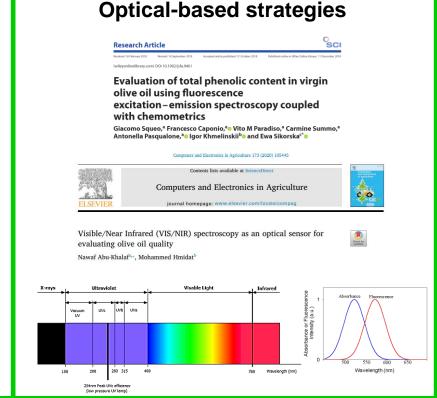








Electrochemical-based strategies



Phenolic content evaluation in Extra Virgin Olive Oil



Why the extraction is required?

COMPOSIZIONE CHIMICA DELL'OLIO EXTRAVERGINE DI OLIVA

L'olio extravergine di oliva è costituito da:

98% gliceridi e acidi grassi monoinsaturi

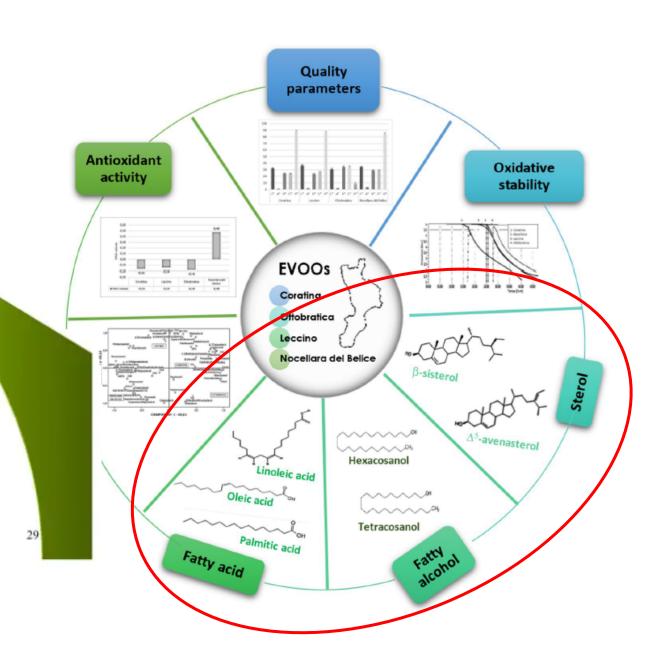
(oleico, linoleico, linolenico)

$$H_{0}C \xrightarrow{H_{2}} H_{2} \xrightarrow{H_{2}} H_{$$

2% componenti minori

(polifenoli, vitamine e sostanze minerali)

05/03/13

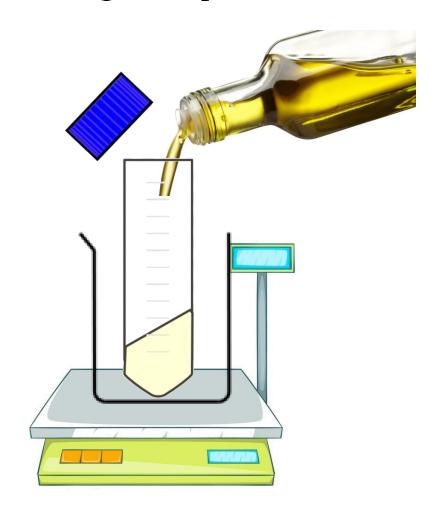


Phenolic content evaluation in Extra Virgin Olive Oil. Main strategies

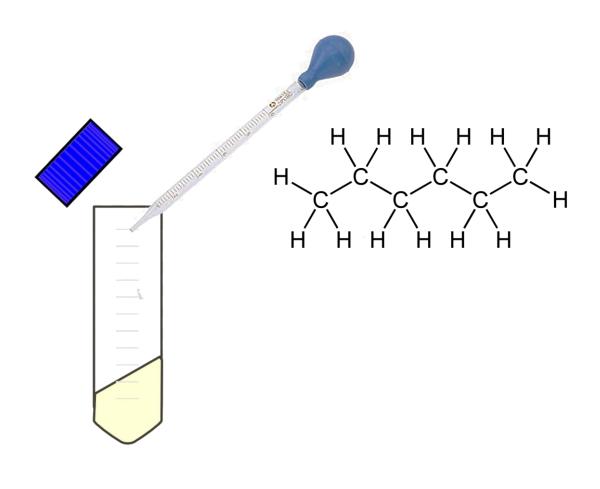


Preparazione del campione

Pesare 1.0 g di campione



Scioglierlo in 5 ml di esano.



Solid phase extraction (SPE). Extraction form liquid sample.

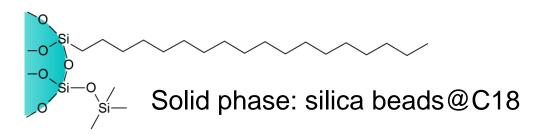
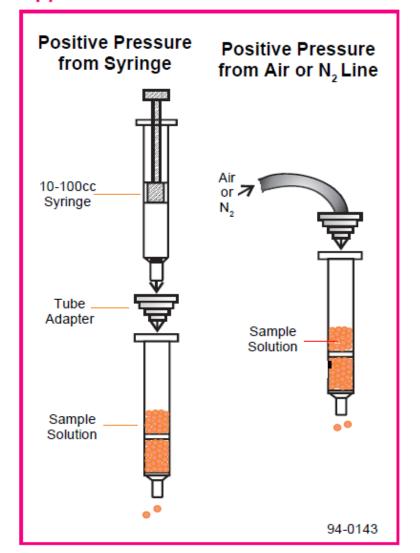




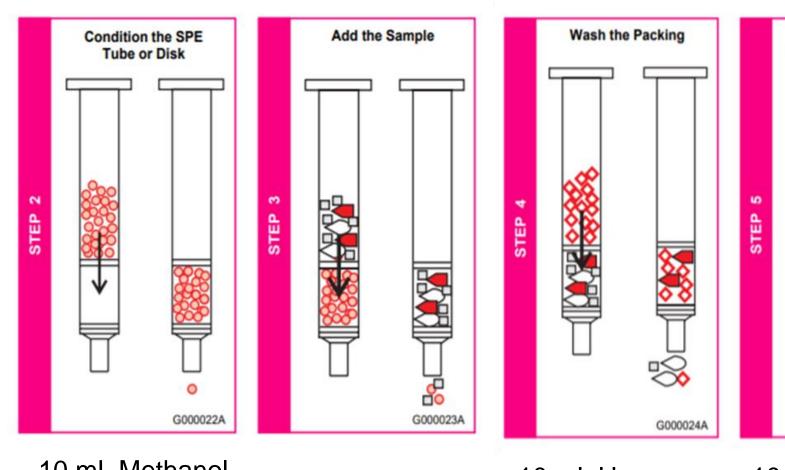
Figure B. Process Using Applied Pressure



Polyphenols extraction

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Solid phase extraction (SPE). Extraction form liquid sample.



Elute the Compounds of Interest G000025A

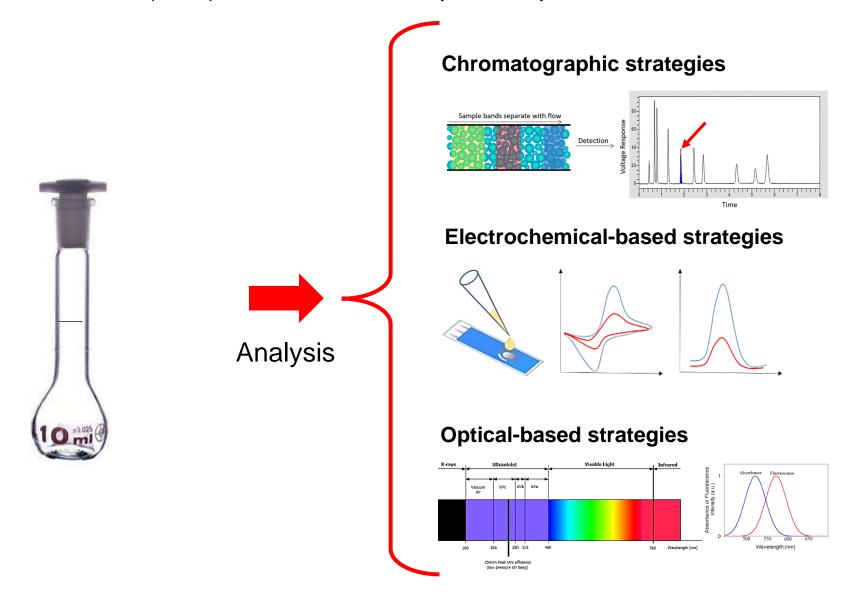
10 mL Methanol 10 mL Hexane

10 mL Hexane

10 mL Methanol

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Solid phase extraction (SPE). Extraction form liquid sample.



Categoria oli Etil esteri degli acidi grassi (EEAG)	ACIDITA' (%)		SPETTROFOTOMETRIA		Valutazione organolettica	Valutazione organolettica		
			K232	K270	ΔΚ	Mediana del difetto (Md)	Mediana del fruttato (Mf)	
VERGINE EXTRA	EEAG ≤ 35 mg/kg (campagna 2014-2016)	≤0,8	≤20	≤2,5	≤0,22	≤0,01	Md = 0	Mf > 0
VERGINE	_	0,8-2	≤20	≤2,6	≤0,25	≤0,01	Md ≤ 3,5	Mf > 0
LAMPANTE	_	> 2	> 20	> 2,6	> 0,25	> 0,01	Md > 3,5	_